

SYLLABUS

Name: Second level of chromatography (WTCCXCSM-SLoCh)

Name in Polish:

Name in English: Second level of chromatography

Information on course:

Course offered by department: Faculty of Advanced Technologies and Chemistry

Course for department: Faculty of Advanced Technologies and Chemistry

Term: Summer semester 2024/2025 Year

Cordinator of course edition: prof. dr hab. inż. Zygfryd Witkiewicz

Default type of course examination report:

Graded pass

Language:

English

Course homepage:

<http://www.wtc.wat.edu.pl>

Short description:

The invention and history of chromatography. Gas, liquid and supercritical chromatography. Capillary electrophoresis. The significance and applications of chromatography.

Description:

1. History of chromatography and its contemporary significance - 1 h .
2. Theory of chromatographic process - 1 h.
3. Gas chromatography, instrumentation, columns, stationary phases, detectors, retention parametrs, column efficiency, qualitative and quantitative analysis - 2 h.
4. Liquid column chromatography, instrumentation, columns, mobile phases, stationary phases, detectors - 2 h.
5. Hyphenated techniques - 1 h..
6. Thin layer chromatography, instrumentation, chromatographic plates. chromatographic chambers, development of chromatograms, visualisation of chromatograms, qualitative and quantitative analysis - 1 h.
7. Supercritical fluid chromatography - 2 h.
7. Biochromatography - 2 h.
8. Capillary electrophoresis, instrumentation and application of this analytical technique 2 h.

Bibliography:

Z. Witkiewicz, J. Kałużna-Czaplińska, Podstawy chromatografii i technik elektromigracyjnych, WNT, Warszawa 2018

Reverse phase HPLC, Macherey Nagel, 2012

W. Jennings, Analytical gas chromatography, Academic Press, 2002

Sz. Nyiredy, Planar chromatography, Springer, 2011

The LC handbook, Agilent Technologies, 2015

Learning outcomes:

- W1 / The student has extended knowledge of analytical chemistry of the selection of the analytical method, allowing for theoretical justification of the choice of the analytical method, determination of the chemical composition of substances or their mixtures / K_W11
- W2 / The student knows classical and instrumental analytical methods, their capabilities and theoretical foundations. He knows the methods of checking the reliability of the results of quantitative chemical analysis and using statistical methods of evaluating the results of the analysis. He knows the tendencies of the development of analytical equipment /K_W12 .
- U1. The student is able to use scientific and research equipment to analyze mixtures and environmental samples /K_U06 .
- U2. The student is able to find the necessary information in literature, databases and other sources, knows the basic scientific journals in the field of chemistry and has the ability to evaluate the information obtained /K_U10.
- U3 / The student is able to independently plan and implement lifelong learning and help others in this area / K_U14.
- U4 / The student is able to interact with others in teamwork and take a leading role in teams/ K_U16
- K1 / The student recognizes the importance of knowledge in solving problems and consulting experts in case of difficulties with self-solving problems / K_K01.
- K2 /The student is ready to fulfill social obligations, inspire and organize activities for the benefit of the social environment and initiate activities for the benefit of social interest / K_K02.

Assessment methods and assessment criteria:

The student should know problems which are delivered during lectures and, in some extent, got during unassisted studies. The student, after passing her or his seminars and laboratory exercises, is allowed to get credit. The credit is oral and covers the knowledge connected with chromatography which has been delivered and discussed during lectures, seminars and laboratory exercises.

The effects W1 and W2 are checked when receiving credit.

The effect U1 is verified during laboratory eXercises and effect U2 during seminars.

Introductory subjects

Analytical chemistry

Physical chemistry

Instrumental analytical chemistry

Programs

Chemistry

Form of course / number of hours / final requirement

Lectures: 14 hours / pass with grade

laboratory exercises: 8 hours / pass without grade

seminar: 8 hours / pass without grade

Author
Prof. dr hab. inż. Zygfryd Witkiewicz
ECTS balance
Activity/load 1. Participation in lectures / 14 hours 2. Independent study of lecture topics / 10 hours 3. Participation in laboratories / 8 hours 4. Independent preparation for laboratories / 6 hours 5. Participation in seminars / 8 hours 6. Independent preparation for seminars / 6 hours 7. Participation in consultations / 2 hours 8. Preparation for passing the course / 6 hours Hours / ECTS Total student workload: 60 hours. /2 Classes with teacher participation: 32 hours / 1 Activities related to scientific activity: 60 / 2
Information on course edition:
Default type of course examination report:
Graded pass
Bibliography:
<i>missing bibliography in English</i>